# NOVEL CLEISTANTHANE DITERPENOIDS FROM POGOSTEMON AURICULARIS ${ }^{1}$ 

Falak A. Hussaini, Sunita Agarwal, Raja Roy, Om Prakash, and Aboo Shoeb*<br>Central Drag Researib Institute. Lnkenuw 226001. India<br>Abstract.-Three novel diterpenoid acids isolated from the whole plant of Pugostemon aurictlaris have been characterized as cleistanth-13,15-dien-18-oic acid and 7-hydroxy- and 7-acetoxycleistanth-13,15-dien-18-oic acids.

Our continued search for new therapeutic agents from medicinal plants endemic to India ( $1-3$ ) led to the examination of Pogostemon auricularis Hassk (Lamiaceae) whose EtOH extract exhibited spasmolytic activity on preliminary screening. From the bioactive residue, obtained from the hexane-soluble extract of the whole plant, three new diterpenes have been isolated. Their structure elucidation is discussed in the present communication.

The first terpene 1, designated as auricularic acid, $\mathrm{mp} 220^{\circ}$, exhibited the presence of an exocyclic methylene group ( $v \max 3070,1644,890 \mathrm{~cm}^{-1} ; \delta 4.57,4.65$ ), a free carboxylic group ( $v \max 3380,1680 \mathrm{~cm}^{-1}$ ), and a vinylic side chain ( $\delta 5.02,5.04$, 6.0 ). In addition, it contained two methyl groups situated in similar environments and resonating almost at the same frequency ( $\delta 1.22,0.71$ ) as $\mathrm{C}-18$ and $\mathrm{C}-20$ methyl groups in gummiferolic acid (4). With these functionalities, the molecular formula, $\mathrm{C}_{20} \mathrm{H}_{30} \mathrm{O}_{2},[\mathrm{M}]^{+}$at $\mathrm{m} / \mathrm{z} 302.2353$, suggested 1 to be a tricyclic diterpene. Compound 1 was converted to the corresponding methyl ester 2 ( $\delta 3.48$, COOMe) with ethereal $\mathrm{CH}_{2} \mathrm{~N}_{2}$.

The dominant feature in the mass spectra of $\mathbf{1}$ and $\mathbf{2}$ was the presence of a conspicuous fragment ion at $m / z 94$ arising from the rupture of ring $C$ and, thus, locating the exocyclic methylene and the vinylic groups at any two of C-11, C-12, C-13, or C-14. The decoupling experiments on $\mathbf{1}$ indicated the vinylic side chain to be situated at a secondary carbon atom that had a methine proton on one side and a quaternary carbon (exomethylene) on the other. For example, irradiation at the frequency of the vinylic methine ( $\delta 6.0$ ) simplified the multiplicities of the vinylic methylene ( $\delta 5.02,5.04$ ), while the allylic methine ( $\delta 2.82 \mathrm{dd}, J=9.2$ and 4.6 Hz ) was converted into a doublet ( $J=4.6 \mathrm{~Hz}$ ). Further, the irradiation at $\delta 2.82$ affected the complexities of a methine ( $\delta 1.50$ ) and those of vinylic methine. Conversely, irradiation at $\delta 1.50$ simplified the multiplicity of the allylic methine to a doublet ( $J=9.2 \mathrm{~Hz}$ ).

The conclusion that the exocyclic methylene and the vinylic side chain were vicinal was further reinforced by the ozonolysis of $\mathbf{1}$. This afforded an aldehyde $\mathbf{3}$ in low yield with $[\mathrm{M}]^{+}$at $m / z 304$, consistent with a composition $\mathrm{C}_{19} \mathrm{H}_{28} \mathrm{O}_{3}$, and a ketone 4 as the major product, which was characterized by its spectral data and its conversion to the corresponding alcohol 5 and acetate 6 . However, the keto aldehyde 7 was never found as a product of the ozonolysis. Besides steric considerations, the difference in the reactivities of the two sites towards ozone may be attributed partly to the difference in their nucleophilicity; the formation of the ozonide of the exocyclic double bond is favored in comparison to the less substituted vinylic double bond. A base peak at $m / z 95.0498$ $\left[\mathrm{C}_{6} \mathrm{H}_{7} \mathrm{O}\right]^{+}$in the mass spectrum of $\mathbf{4}$ representing the fragment ion $\mathbf{8}$ ( 5 ) corroborates the vicinal disposition of the above two functionalities and implies the preferred placement of the exocyclic methylene and the vinylic side chain at $\mathrm{C}-13$ and C -14, respectively.

Further evidence for the structure $\mathbf{1}$ for auricularic acid was found by a ${ }^{1} \mathrm{H}-\mathrm{nmr}$ study of the aldehyde 3. Irradiation at $\delta 9.79(\mathrm{~d}, J=9 \mathrm{~Hz}, \mathrm{CHO}$ ) converted $\mathrm{H}-14$ (dd, $J=9$ and 4.5 Hz ) into a doublet ( $J=4.5 \mathrm{~Hz}$ ).

The structure $\mathbf{1}$ was finally confirmed by the dehydrogenation of auricularic acid over $\mathrm{Pd} / \mathrm{C}$, which gave two products. The less polar of the two, mp $108-109^{\circ}$, was identified as 1,7 -dimethyl-8-ethylphenanthrene; the other, $\mathrm{mp} 141^{\circ}$, was characterized as cleistanth-8,11,13-trien-18-oic acid [9]. These two new products confirmed the location of the exomethylene and the vinylic groups in 1 at $\mathrm{C}-13$ and C -14, respectively. Compound 1 may thus be formulated as cleistanth-13,15-dien-18-oic acid.

The complete stereostructure of $\mathbf{1}$ has been determined with the help of $2 \mathrm{D}{ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ (COSY, NOESY) and ${ }^{1} \mathrm{H}_{-}^{13} \mathrm{C}$ (Hetero COSY) nmr studies and is reported separately (6).

The second terpene 10, an amorphous solid, $[\mathrm{M}]^{+}$at $\mathrm{m} / \mathrm{z} 318\left(\mathrm{C}_{20} \mathrm{H}_{30} \mathrm{O}_{3}\right)$, gave an acetate, $\mathrm{mp} 178^{\circ},[\mathrm{M}]^{+}$at $m / z 360$, which was identical in all respects to the third terpene 11 isolated as a natural product. Possessing functionalities similar to $\mathbf{1}$ and in a similar environment, $\mathbf{1 0}$ may be considered as a hydroxy derivative of $\mathbf{1}$. This view was further reinforced by Jones's oxidation of $\mathbf{1 0}$ to yield the ketone $\mathbf{1 2}, \mathrm{mp} 154^{\circ},[\mathbf{M}\}^{+}$at $m / z 316\left(\mathrm{C}_{20} \mathrm{H}_{28} \mathrm{O}_{3}\right)$, which under modified Wolff-Kishner reduction conditions (7) gave auricularic acid $\{\mathbf{1}\}$ (Scheme 1).


|  | $\mathrm{R}_{1}$ | $\mathrm{R}_{2}$ | $\mathrm{R}_{;}$ |
| ---: | :--- | :--- | :--- |
| $\mathbf{1}$ | H | $=\mathrm{CH}_{2}$ | H |
| $\mathbf{2}$ | Me | $=\mathrm{CH}_{2}$ | H |
| $\mathbf{4}$ | H | $\mathbf{O} \mathrm{O}_{2}$ | H |
| $\mathbf{5}$ | H | $\mathrm{H}, \mathrm{OH}_{2}$ | H |
| $\mathbf{6}$ | H | $\mathrm{H}, \mathrm{OAc}$ | H |
| $\mathbf{1 0}$ | H | $=\mathrm{CH}_{2}$ | OH |
| $\mathbf{1 1}$ | H | $=\mathrm{CH}_{2}$ | OAc |
| $\mathbf{1 3}$ | Me | $=\mathrm{CH}_{2}$ | OH |
| $\mathbf{1 4}$ | Me | $=\mathrm{CH}_{2}$ | OAc |



$$
\begin{array}{ll}
3 & \mathrm{R}=\mathrm{CH}_{2} \\
7 \mathrm{R} & =\mathrm{O}
\end{array}
$$



10
12
1

$m / z 154$

m/z 162 (100\%)

Scheme 1

The hydroxy group of $\mathbf{1 0}$ may be placed on any one of the three rings of $\mathbf{1}$. However, its presence in ring $C$ is ruled out on the basis of the study of the mass spectrum of 10 that showed conspicuous fragment ions at $m / z 94,146,164,193$, and 206 arising from the rupture of ring C . This, together with the observed deshielding of $\mathrm{H}-14$ in $\mathbf{1 0}$ and 11 and their respective methyl esters 13 and 14 and in 12 compared to 1 suggests the presence of an oxygen function in ring B , probably at $\mathrm{C}-7$. Another piece of evidence in support of this view was provided by the observation of two conspicuous fragment ions at $m / z 154$ and 162 (base peak) in the mass spectrum of $\mathbf{1 2}$ resulting from the rupture of ring B in the manner depicted in Scheme 1.

The final proof in favor of structure $\mathbf{1 0}$ was achieved by the study of $400 \mathrm{MHz}{ }^{1} \mathrm{H}$ nmr spectra of 10 and 11. The $\mathrm{H}-7$ located at $\delta 4.52$ in 11 [ddd, $J_{7,6 \mathrm{eq}}=4.6$; $J_{7, \text { Gax }}=11.0$ and $J_{7,8 a x}=12.1 \mathrm{~Hz}$ \} was confirmed by specific proton decoupling technique. Thus, for example, irradiation at $\delta 4.52$ caused an enhancement in the $\mathrm{C}-7$ signal resonating at $\delta 77.0$ in ${ }^{13} \mathrm{C}$-nmr spectrum of 11. These, together with the broad band (BB) and single frequency off-resonance decoupling methods in their ${ }^{1.3} \mathrm{C}-\mathrm{nmr}$ spectra, permitted the assignment of all the carbon atoms in $\mathbf{1 0}$ and 11. Further, the multiplicity associated with $\mathrm{H}-7$ mentioned above conclusively demonstrated $\mathrm{C}-7$ as the locus of equatorial hydroxy and acetoxy groups in 10 and 11 , respectively.

Compound 11 inhibited 10,19 , and $69 \%$ response of spasmogens at $2.5,5$, and 10 $\mu \mathrm{g} / \mathrm{ml}$, respectively, on guinea pig ileum (8).

## EXPERIMENTAL

Mass spectra were obtained on a JEOL D-300 instrument with a JMA-2000 data system. All compounds gave satisfactory elemental analyses.

Plant material. - The whole plant of $P$. amrichlarij was collected in the month of February from Thalacauvery, Karnatak, India. A voucher specimen has been deposited in the herbarium of the Borany Section of the Institute.

Extraction.-Dried and finely powdered plant material ( 3 kg ) was extracted with hexane $(5 \times 5$ liters). The extract was concentrated in vacuo to leave a residue ( 25 g ) that was separated into acidic and nonacidic fractions by treatment with $1 \% \mathrm{NaOH}$ solution. The basic solution on acidification followed by extraction with $\mathrm{Et}_{2} \mathrm{O}$ gave the acidic fraction ( 20.2 g ). The nonacidic fraction ( 3.3 g ) on cc ( Si gel ) afforded methyl linolenate and sitosterol. The acidic fraction on similar treatment yielded a mixture of diterpenoid acids $(3.0 \mathrm{~g})$ from which $\mathbf{1}(0.75 \mathrm{~g}), \mathbf{1 0}(0.15 \mathrm{~g})$, and $\mathbf{1 1}(0.27 \mathrm{~g})$ were obtained after rechromatography ( Si gel; $\mathrm{AgNO}_{3}$ ) of the mixture.

Auricularic acid $[\mathbf{1}\}-\mathrm{Mp} 220^{\circ}(\mathrm{MeOH}) ;[\alpha]^{25} \mathrm{D}+15.2^{\circ}\left(c=2, \mathrm{CHCl}_{3}\right)$; ir $v \max (\mathrm{KBr}) 3380,3070$, $2930,1680,1644,1450,1232,1226,918,890 \mathrm{~cm}^{-1} ; \mathrm{ms} \mathrm{m} / \mathrm{z}[\mathrm{M}]^{+} 302.2353,287,257,241,207$, $149,148,135,1.34,120,94,87,80$.

Metbyl ester 2.—Amorphous solid; ir $v \max (\mathrm{KBr}) 3450,2950,1738,1458,1232,1190,1165$, $1105,1000,925,900,838,790 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}(90 \mathrm{MHz}) \delta 5.95(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-15), 4.96(1 \mathrm{H}, \mathrm{dd}, J=10$ and $1.0 \mathrm{~Hz}, \mathrm{H}-16), 4.90\left(1 \mathrm{H}, \mathrm{dd}, J=18\right.$ and $\left.1.0 \mathrm{~Hz}, \mathrm{H}-16^{\prime}\right), 4.50,4.43(1 \mathrm{Heach}, \mathrm{q}, J=2 \mathrm{~Hz}, 2 \mathrm{H}-17)$, $3.48\left(3 \mathrm{H}, \mathrm{s}, \mathrm{COOCH}_{3}\right), 2.68(1 \mathrm{H}, \mathrm{dd}, J=9$ and $4.5 \mathrm{~Hz}, \mathrm{H}-14), 1.2(\mathrm{~s}, 19-\mathrm{Me}), 0.52(\mathrm{~s}, 20-\mathrm{Me}) ; \mathrm{ms} \mathrm{m} / \mathrm{z}$ ${ }_{[M]}{ }^{\text {t }} 316,301,257,241,222,162,161,148,147,134,121,101,94$.

OZONOLYSIS OF 1 TO YIELD 3 AND 4.-Treatment of a solution of $\mathbf{1}$ ( 200 mg ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 100 ml ) at $-70^{\circ}$ with a current of ozone for 4 h followed by usual work-up afforded 3 and 4 in 5 and $50 \%$ yields, respectively.

14-Formylcleistantb-13-en-18-vic acid [3]. $-\mathrm{Mp} 170^{\circ}$; ir $v \max (\mathrm{KBr}) 2975,1740,1700,1460$, 1280, 1262, 1182, 1160, 1150, $980,913,912 \mathrm{~cm}^{-1} ;{ }^{'} \mathrm{H} \mathrm{nmr}(90 \mathrm{MHz}) \delta 9.79(1 \mathrm{H}, \mathrm{d}, J=9 \mathrm{~Hz}, \mathrm{CHO})$, 4.96 and $4.81(1 \mathrm{H}$ each, $\mathrm{q}, J=2 \mathrm{~Hz}, 2 \mathrm{H}-17), 3.13(1 \mathrm{H}, \mathrm{dd}, J=9$ and $4.5 \mathrm{~Hz}, \mathrm{H}-14), 1.26$ (s, 19-Me), 0.73 (s, 20-Me); ms m/z[M] ${ }^{+} 304,208,276,275,260,208,194,150,135,134,121,96,66$

13-Oxocileistanth-15-en-18-oic acid [4].—Mp 153 ${ }^{\circ}$; ir $v \max (\mathrm{KBr}) 3100,2900,1732,1702,1452$, 1393, 1329, 1250, 1220, $1160,998,925 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}(90 \mathrm{MHz}) \delta 5.90(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-15), 5.15(1 \mathrm{H}, \mathrm{dd}$, $J=10$ and $1.0 \mathrm{~Hz}, \mathrm{H}-15), 5.08(1 \mathrm{H}, \mathrm{dd}, J=18$ and $1.0 \mathrm{~Hz}, \mathrm{H}-16), 2.90(1 \mathrm{H}, \mathrm{dd}, J=9$ and $4.5 \mathrm{~Hz}, \mathrm{H}-$ 14), 1.12 (s, 19-Me), and $0.7(\mathrm{~s}, 20-\mathrm{Me})$; ms m/z [M] ${ }^{\dagger} .304,208,193,148,139,135,95.0498$ $\left[\mathrm{C}_{6} \mathrm{H}_{7} \mathrm{O}\right]^{+}$base peak.
$\mathrm{NaBH}_{4}$ Reduction of 4 to yield 13-hydroxycleistanth-15-en-18-oic acid [5].-NaBH was slowly added to a cold MeOH solution of $\mathbf{4}(100 \mathrm{mg})$, and the mixture stirred for 4 h . The reaction product on work-up afforded $5, \mathrm{mp} 189$; ir $\nu \max (\mathrm{KBr}) 3450,2950,1688,1463,1060,920,801 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}(90 \mathrm{MHz}) \delta 5.90(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-15), 5.23(1 \mathrm{H}, \mathrm{dd}, J=10$ and $1.0 \mathrm{~Hz}, \mathrm{H}-16), 5.11(1 \mathrm{H}, \mathrm{dd}, J=18$ and $1.0 \mathrm{~Hz}, \mathrm{H}-16), 3.56(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-13$ and $\mathrm{H}-14), 1.25(\mathrm{~s}, 19-\mathrm{Me}), 0.74(\mathrm{~s}, 20-\mathrm{Me}) ; \mathrm{ms} m / \approx[\mathrm{M}]{ }^{+} 306$, 291, 288, 278, 261, 220, 208, 80.

Acetylation of 5 to yield 13-acetoxycleistanth-15-en-18-oic acid [6].-Compound 5 on treatment with $\mathrm{Ac}_{2} \mathrm{O} /$ pyridine yielded 6 , $\mathrm{mp} 199-200^{\circ}$; ir $v \max (\mathrm{KBr}) 3500,2950,1750,1708$, $1459,1377,1247,1120,1043,981,924 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}(90 \mathrm{MHz}) 85.90(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-15), 5.1(1 \mathrm{H}, \mathrm{dd}$, $J=10$ and $1.0 \mathrm{~Hz}, \mathrm{H}-16), 5.09(1 \mathrm{H}, \mathrm{dd}, J=18$ and $1.0 \mathrm{~Hz}, \mathrm{H}-16), 4.75(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-13), 2.68(1 \mathrm{H}, \mathrm{m}$, H-14), $2.02(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}), 1.36(\mathrm{~s}, 19-\mathrm{Me}), 0.79(\mathrm{~s}, 20-\mathrm{Me}) ; \mathrm{ms} m / 2[\mathrm{M}] .348,288,287,273,272,258$, $242,172,135,120$.

Dehydrogenation of 1.-Compound $1(100 \mathrm{mg})$ was heated with $10 \% \mathrm{Pd} / \mathrm{C}(200 \mathrm{mg})$ at $300^{\circ}$ in $\mathrm{N}_{2}$ atmosphere for 10 min and repeatedly extracted with $\mathrm{CHCl}_{3}$. After removal of the solvent, the residue on cc (Si gel) gave the following two products.
1.7-Dimethyl-8-ethylphenanthrene.-Mp 108-109 ; ir $v \max (\mathrm{KBr}) 3070,3020,2970,2937,2875$, 1595, 1452, 1378, 1309, 1255, 1210, 1182, 1072, 1040, 828, $805,771 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}(400 \mathrm{MHz}) \delta$ $8.49(1 \mathrm{H}, \mathrm{d}, J=8 \mathrm{~Hz}, \mathrm{H}-4), 8.42(1 \mathrm{H}, \mathrm{d}, J=8.6 \mathrm{~Hz}, \mathrm{H}-5), 7.94(2 \mathrm{H}, \mathrm{ABq}, \mathrm{A}=7.90, \mathrm{~B}=7.99, J=9.4$ $\mathrm{Hz}, \mathrm{H}-9$ and $\mathrm{H}-10), 7.44(1 \mathrm{H}, \mathrm{t}, J=8 \mathrm{~Hz}, \mathrm{H}-3), 7.39(1 \mathrm{H}, \mathrm{d}, J=8.6 \mathrm{~Hz}, \mathrm{H}-6), 7.34(1 \mathrm{H}, \mathrm{d}, J=8 \mathrm{~Hz}$, $\mathrm{H}-2), 3.09\left(2 \mathrm{H}, \mathrm{q}, J=7.5 \mathrm{~Hz}, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 2.69(3 \mathrm{H}, \mathrm{s}, 1-\mathrm{Me}), 2.48(3 \mathrm{H}, \mathrm{s}, 7-\mathrm{Me}), 1.23(3 \mathrm{H}, \mathrm{r}, J=7.5$ $\left.\mathrm{Hz}, \mathrm{CH}_{3} \mathrm{CH}_{2}\right) ; \mathrm{ms} m / z[\mathrm{M}] \div 234,219,190,101,89,85,83$.

Cleistanth-8.11.13-trien-18-aic acid [9]. $-\mathrm{Mp} 141^{\circ}$; ir $\nu \max (\mathrm{KBr}) 2900,1670,1460,1440,1400$, $1370,1320,1305,1218,1200,1155,1100,1060,1030,982,930,910,815,800,740,665,620$
$\mathrm{cm}^{-1}$; ${ }^{\mathrm{t}} \mathrm{H} \mathrm{nmr}(90 \mathrm{MHz}) \delta 6.98(1 \mathrm{H}, \mathrm{d}, J=9 \mathrm{~Hz}, \mathrm{H}-11), 6.85(1 \mathrm{H}, \mathrm{d}, J=9 \mathrm{~Hz}, \mathrm{H}-12), 2.58(2 \mathrm{H}, \mathrm{q}$, $\left.J=7 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 2.22(3 \mathrm{H}, \mathrm{s}, \mathrm{ArCH} ;), 1.24(3 \mathrm{H}, \mathrm{s}, 20-\mathrm{Me}), 1.21(3 \mathrm{H}, \mathrm{s}, 19-\mathrm{Me}), 1.04(3 \mathrm{H}, \tau, J=7$ $\left.\mathrm{Hz}, \mathrm{CH}_{2} \mathrm{CH}_{i}\right) ; \mathrm{ms} m / z[\mathrm{M}]+300,285(100 \%), 239,183,159,157,85,71,57,55$

7-Hydroxydeistanth-13,15-dien-18-uic acid [10]. -Ir $v \max (\mathrm{KBr}) 3440,2950,2908,1700,1660$, $1450,1398,1239,1180,1018,910,840 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}(400 \mathrm{MHz}) \delta 1.12\left(J_{5,(\mathrm{xq}}=2.4, J_{5,6 \mathrm{ax}}=13.1\right.$, $\mathrm{H}-5), 1.82(\mathrm{~m}, \mathrm{H}-6 \mathrm{ax}), 2.13(\mathrm{~m}, \mathrm{H}-6 \mathrm{eq}), 3.40\left(J_{7 . \text { xuy }}=4.6, J_{7 . \text { (ax }}=11.0, J_{7 . \mathrm{s}}=12.1, \mathrm{H}-7\right), 1.48(\mathrm{~m}$, $\left.\mathrm{H}-8), 1.11(\mathrm{~m}, \mathrm{H}-9), 3.39\left(J_{14.8}=4.6\right), J_{1+.15}=9.8, \mathrm{H}-14\right), 6.06\left(J_{15.14}=9.2, J_{15.16}=9.8\right.$, $\left.J_{15.16}=17.3, \mathrm{H}-15\right), 5.09\left(J_{k * m}=1.7, J_{16.15}=9.8, \mathrm{H}-16\right), 5.20\left(J_{\text {gewn }}=1.7, J_{16.15}=17.3, \mathrm{H}-16^{\prime}\right)$, $4.73,4.61\left(J_{17.17}=2.3, \mathrm{H}-17,17^{\prime}\right), 1.24(\mathrm{Me}-19), 0.73(\mathrm{Me}-20) ;{ }^{13} \mathrm{C} \mathrm{nmr}(100.57 \mathrm{MHz}) \delta 31.2(\mathrm{C}-1)$, $19.4(\mathrm{C}-2), 39.4(\mathrm{C}-3), 43.6(\mathrm{C}-4), 52.8(\mathrm{C}-5), 32.3(\mathrm{C}-6), 72.0(\mathrm{C}-7), 46.7(\mathrm{C}-8), 48.1$ (C-9), 37.3 (C10), 27.1 (C-11), $37.8(\mathrm{C}-12), 150.8(\mathrm{C}-13), 49.1(\mathrm{C}-14), 136.5(\mathrm{C}-15), 116.6(\mathrm{C}-16), 107.4(\mathrm{C}-17)$, 182.9(C-18), 28.9(C-19), 12.8(C-20); ms m/z318, 300, 236, 224, 223, 206, 194, 193, 164, 146, 109, 95, 94.

7-Acetwxyleistanth-13.15-dien- 18 -aic acid [11].-Compound $\mathbf{1 0}$ ( 50 mg ) on treatment with $\mathrm{Ac}_{2} \mathrm{O} /$ pyridine gave $11, \mathrm{mp} 178^{\circ}$, identical in all respects to the natural product, ir $\nu \max (\mathrm{KBr}) 3320,2908$, $1724,1660,1450,1380,1250,1038,980,940 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}(400 \mathrm{MHz}) \delta 2.19(\mathrm{~m}, \mathrm{H}-1), 1.48(\mathrm{~m}, \mathrm{H}-$ 2ax), 1.71 (m, H-2eq), 1.00 (m, H-3ax), 1.78 ( $\mathrm{m}, \mathrm{H}-3 \mathrm{eq}$ ), 1.19 ( $J_{5, \text { (xeq }}=2.3, J_{5, \text { fux }}=13.2, \mathrm{H}-5$ ), 1.75 ( $\mathrm{m}, \mathrm{H}-\mathrm{Gax}$ ), $2.26\left(J_{6.5}=2.3, J_{6.5}=4.6, J_{\text {ргm }}=12.5, \mathrm{H}-6 \mathrm{eq}\right), 4.52\left(J_{7 . \text { (к.4 }}=4.6, J_{7 . \text { oux }}=11.0\right.$, $\left.J_{7.8}=12.1, \mathrm{H}-7\right), 1.72(\mathrm{~m}, \mathrm{H}-8), 1.15(\mathrm{~m}, \mathrm{H}-9), 1.81(\mathrm{~m}, \mathrm{H}-1 \mathrm{lax}), 1.88(\mathrm{~m}, \mathrm{H}-1 \mathrm{leq}), 1.14$ (m, H$12 \mathrm{ax}), 2.17(\mathrm{~m}, \mathrm{H}-12 \mathrm{eq}), 3.16\left(J_{11.8}=4.6, J_{11.15}=9.2, \mathrm{H}-14\right), 6.00\left(J_{15.11}=9.2, J_{15.16}=9.8\right.$, $\left.J_{15.16}=17.3, \mathrm{H}-15\right), 5.05\left(J_{\mathrm{gem}}=1.7, J_{16.15}=9.8, \mathrm{H}-16\right), 4.96\left(J_{\mathrm{g} \mathrm{cm}}=1.7, J_{16.15}=17.3, \mathrm{H}-16^{\prime}\right)$, $4.73,4.63\left(J_{1-.1}=2.3, \mathrm{H}-17,17^{\prime}\right), 1.23(\mathrm{Me}-19), 0.73(\mathrm{Me}-20), 2.07\left(\mathrm{CH}_{3} \mathrm{CO}\right) ;{ }^{19} \mathrm{C} \mathrm{nmr}(100.57$ MHz ) $\delta 31.0(\mathrm{C}-1), 19.3(\mathrm{C}-2), 39.3(\mathrm{C}-3), 43.7(\mathrm{C}-4), 52.3(\mathrm{C}-5), 28.4(\mathrm{C}-6), 77.0(\mathrm{C}-7), 44.9(\mathrm{C}-8)$, 46.8(C-9), 37.2(C-10), 27.1(C-11), $37.6(\mathrm{C}-12), 150.7(\mathrm{C}-13), 49.2$ (C-14), 136.1(C-15), 116.8(C16), $107.7(\mathrm{C}-17), 183.0(\mathrm{C}-18), 12.7(\mathrm{C}-20), 170.3(\mathrm{CH} ; \mathrm{CO}), 21.0(\mathrm{CH} ; \mathrm{CO}) ; \mathrm{ms} m / z[\mathrm{M}-60]^{+} 300$, $285,205,146,133,109,94,92,91$.

7-Oxwleistanth-13.15-dien-18-nic acid [12].—Excess Jones's reagent was added to a solution of 10 ( 100 mg ) in $\mathrm{Me}_{2} \mathrm{CO}$ and the mixture left overnight. Cc of the product afforded 12, $60 \mathrm{mg}, \mathrm{mp} 154^{\circ} ; \delta$ ( 90 MHz ), 3.61 ( $\mathrm{m}, \mathrm{H}-14$ ), $5.76\left(\mathrm{~m}, \mathrm{H}-15\right.$ ), 5.14 (dd, $J_{\text {grm }}=1.2, J_{16.15}=10.3, \mathrm{H}-16$ cij), 4.98 (dd, $J_{s t / m}=1.2, J_{16,15}=17.3, \mathrm{H}-16$ trans), 4.67 (bs, H-17,17'), 1.19 (s, Me-19), 0.84 (s, Me-20); ir $v$ max (KBr) $3300,2980,2910,1720,1700,1660,1470,1450,1300,1250,1115,908 \mathrm{~cm}^{-1} ; \mathrm{ms} m / z 316$, 270, 227, 162 (100\%), 154, 149, 144, 121, 119, 109, 107, 105, 93.

Methylation of 10 .-Compound 10 ( 100 mg ) was methylated with $\mathrm{CH}_{2} \mathrm{~N}_{2}$ in $\mathrm{Et}_{2} \mathrm{O}$ to give 13 $(95 \mathrm{mg}), \delta(90 \mathrm{MHz}) 3.45(\mathrm{~m}, \mathrm{H}-7$ and $\mathrm{H}-14), 6.02(\mathrm{~m}, \mathrm{H}-15), 5.28\left(\mathrm{dd}, J_{\mathrm{sk} \mathrm{m}}=1.2, J_{16.15}=10.3, \mathrm{H}-16\right.$ (ij), $5.20\left(\mathrm{dd}, J_{\text {дems }}=1.2, J_{16.15}=17.3, \mathrm{H}-16 \mathrm{trans}\right), 4.71,4.61\left(\mathrm{q}, J=2.3, \mathrm{H}-17,17^{\prime}\right), 1.17(\mathrm{~s}, \mathrm{Me}-19)$, 0.63 ( $\mathrm{s}, \mathrm{Me}-20$ ), 3.63 ( $\mathrm{s}, \mathrm{COOCH}$; ; ir $v \max$ (neat) $3340,2960,2908,1720,1660,1450,1240,1160$, $910,900 \mathrm{~cm}^{-1} ; \mathrm{ms} \mathrm{m} / z 332,314,299,272,254,238,207,148,146,135,133,109,94,90$.

Methylation of 11 - Compound 11 ( 100 mg ) on methylation with $\mathrm{CH}_{2} \mathrm{~N}_{2}$ in $\mathrm{Et}_{2} \mathrm{O}$ afforded 14 $(95 \mathrm{mg}) ; \delta(90 \mathrm{MHz}) 3.08(\mathrm{dd}, J=4.5,9.0 \mathrm{~Hz}, \mathrm{H}-14), 4.76(\mathrm{~m}, \mathrm{H}-7), 5.90(\mathrm{~m}, \mathrm{H}-15), 4.98$ (dd, $J_{\text {krm }}=1.2, J_{16.15}=10.3 \mathrm{~Hz}, \mathrm{H}-16$ (is), $4.92\left(\mathrm{dd}, J_{\text {grm }}=1.2, J_{16.15}=17.3, \mathrm{H}-16\right.$ tram. $), 4.72,4.52$ (q, $J=2.3, \mathrm{H}-17,17^{\prime}$ ), $1.16(\mathrm{~s}, \mathrm{Me}-19), 0.59(\mathrm{~s}, \mathrm{Me}-20), 3.55\left(\mathrm{~s}, \mathrm{COOCH}_{3}\right), 2.00(\mathrm{~s}, \mathrm{OAc})$; ir $v$ max (neat) 3340, 3000, 2904, 1740, 1640, 1380, 1245, 1220, $1165 \mathrm{~cm}^{-1}$; ms m/z 374, 332, 314, 254, 206, 146, 145, 134, 94, 92.

Wolff-Kishner reiduction of 12 to yielid 1 - Compound $12(80 \mathrm{mg})$ was reduced under $\mathrm{N}_{2}$, in conditions described by Nagata and Itazaki (7) to yield a product which on purification by cc (Si gel) furnished $\mathbf{1}(33 \mathrm{mg})$ identical in all respects (ir, tlc, nmr) with an authentic sample.

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